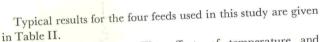


Figure 2. Apparent effect of sample weight on rate of char hydrogasification at 1700° F. and 1500 p.s.i.g.



Effect of Variables. The effects of temperature and extent of conversion on the rate of reaction of low-temperature bituminous coal char and hydrogen were measured in a series of tests conducted at 1500 p.s.i.g. and at 1300°, 1500°, and 1700 ° F. (Figure 4). During the initial phases, the reaction rate was not significantly affected by temperature in the range studied. Only after approximately 20% carbon gasification did the effects of temperature become apparent. The rate constants for the residual char would be expected to follow the pseudo-first-order relationship:

$$r = kp$$

where r = rate of reaction in pounds of carbon in gaseous hydrocarbons per hour per pound of carbon in bed

k = rate constant

p = hydrogen partial pressure in atmospheres

This expression has been shown by Blackwood (2) to be applicable in the temperature range of 650° to 870° C. (1202° to 1598° F.) for the reaction of coconut char with excess hydrogen at pressures up to 40 atm. Birch, Hall, and Urie (1) have also applied it successfully to correlate data on the hydrogenation of the residual (aromatic) carbon portion of Australian brown coal with excess hydrogen in a fluid-bed reactor for the temperature range from 750° to 950° C. (1382° to 1742° F.). Zielke and Gorin (15) showed that, from 1500° to 1700° F. and at 1 to 30 atm. with devolatilized Disco bituminous coal char, the apparent reaction order is 2 at low pressures and approaches 1 at high pressures.

In Table III, pseudo-first-order hydrogasification rate constants for these chars are compared with the values for lowtemperature bituminous coal char after 25 to 30% carbon conversion (Figure 4). Agreement is good, except for the acid-

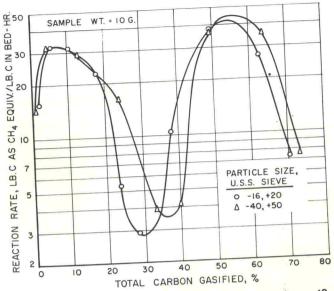


Figure 3. Effect of char particle size on rate of hydrogasification at 1700° F. and 1500 p.s.i.g.

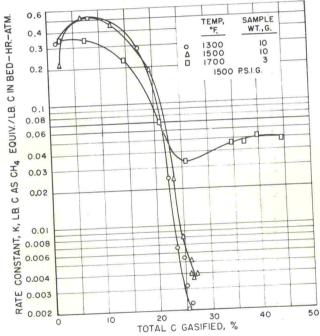


Figure 4. Effect of temperature and conversion on reaction rate constant for bituminous coal char

extracted, high-temperature coconut char. The rates for this specially prepared low-reactivity material are up to one order of magnitude lower, as would be expected.

All of the above results were obtained in differential-bed reactors of various types, except for the data for Australian brown coal, which were obtained in an integral fluid-bed reactor. However, methane concentrations in the product gases were low enough to minimize equilibrium hindrance

Feed	Tabl 5 gr	e II. Ty	pical Te	st Result	s at 170	O° F. and	1500 P.S	i.I.G.			-H-06-a
Time of sampling, sec.	10) 20	hommons	s coal, —	16, +20	U. S. S. si	eve size				
1 emperature ° F	1740	20) 2	.5	30 3	5 4		80	120	240	100
Feed hydrogen rate, SCF/hr.	104									240 1732	480
Exit gas rate, SCF/br	103				04.3 10	4.3 10	4.3 104	- /			1725
Exit gas composition, mole %	100	5.7 102	2 10	1.5 10	01.5 10	0.8 100	0.8 101				103.
1V2 + CO		.05 0	OF	0 51				100.	101.	4 102.7	100.
CO_2				0.51		0.61 (0.49	.12 0.	04 0.0	0.0	
H_2	99	.94 99		0.03		0.01 (0.01			0.04	4 0.0
CH ₄					7.48 8).14 96	25 98.		00.0	
C_2H_6		The same of				9.88		55 1.			
Benzene							0.01	7.0		0.99	0.
Total	100		-					08 0.		1 0 04	
	100	.00 100	.00 100	$0.00 \overline{10}$	0.00 10		0.00 100.			_	
Rate of formation of gaseous hyd	dro-					100	.00 100.	00 100.	00 100.0	00 100.00	100.0
carbon carbon, lb. /lb. carbon											
fed-hr.			20	.2 4							
Total conversion of carbon in fee	ed,		20	. 2 4	5.4 40	1.4 36	.9 15.	4 6.9	9 3.2	4.1	1.6
/0			0	.4	7 0 40				0.2	7.1	1.0
Total carbon recovery, %						.5 17	.5 32.	6 38.4	4 43.4	56.6	77.0
Feed											
Time of sampling, sec.	5 grai	ns of lign	ite, -16	. +20 U	S. S. sie	re cizo			•••		87.9
Temperature, ° F.	10	20	25	30							
Feed hydron F.	1712	1722	1723	1725			80	120	240	480	600
Feed hydrogen rate, SCF/hr.	98.		9 99			1726	1721	1717	1714	1713	1714
Dait gas rate. SCF/hr	98.	7 98.					_			97.6	95.9
Exit gas composition, mole %			100	91	.6 95	.2 98.	3 97.1	97.0		96.6	94.4
1V2 + CO	0.	06 0.0	04 0	.56 1	02	01				20.0	24.4
CO_2							02 0.0	0.0	5 0.04	0.06	0.0
\mathbf{H}_2	99.	93 99					03			0.00	0.0
CH ₄	0.0				.68 88		57 98.7	7 99.3	8 99.58	00.60	00 7
C_2H_6						15 5.					99.7.
Benzene		***			.01			-			0.2
Total	100 (100	_			05 0.	01			*	
	100.0	00 100.0	00 100.	00 100	.00 100.	00 100.		100.00	100.00	1 ***	
Rate of formation of gaseous hydr	0-				100.	00 100.1	100.0	0 100.00	100.00	100.00	100.00
carbon carbon, lb. /lb. carbon											
icu-nr.		0.1	44								
Total conversion of carbon in feed		0.1	11.	6 39.	7 41.	7 24.8	5.2	2.6	1 7	- 1 - 2 - 2	Service.
/0			100	40			0.2	2.0	1.7	1.1	0.9
Total carbon recovery, %			0.8	8 5.	9 20.	0 32.2	42.1	16.0	F0 0	CI BUILD OF	
Feed								46.0	53.3	62.9	66.3
Time of a 1	5 gram	s of media	ım volat	iliter anth			J. S. S. sie		2		82.3
Time of sampling, sec.	10	20	25	mty anti	racite, -	16, +20	J. S. S. sie	ve size			
Temperature, ° F.	1696	1694	1693	50	33	40	60	120	240	360	600
Feed hydrogen rate, SCF/hr.	97.4	97.3		1695	1698	1702	1702	1702	1700	4000	600
Exit gas rate, SCF/hr	96.9	95.8	97.3					97.2	97.0		698
Exit gas composition mole 07	20.7	93.0	95.3	95.	7 93.0	93.1	94.3	95.4	94.5	96.9	97.4
N ₂ + (i()	0.04	0 10						23.4	94.3	95.4	96.1
CO_2	0.01	0.10				0 0.1	7 0.07	0.04	0.02	0.00	
H_2	99.95							0.04	0.03	0.03	0.03
CH ₄						7 94.5		98.90	00 11		
$C_2\dot{H_6}$		0.22			5.8	3 5.2		1.04	99.11	99.15	99.52
C_5H_{12}			0.0	1				1.04	0.86	0.82	0.45
Mono-olefins		***						0.04			
Benzene							0.01	0.01			
Total	10.00		0.0	0.0	1		0.01	0.01	1		
	100.00	100.00	100.00			100.00				4.14	
Rate of formation of gaseous hydro-			-00.00	100.0	0 100.0	100.00	100.00	100.00	100.00	100.00 1	100.00
carbon carbon. Ib. /lb carbon										10.00	50.00
red-nr.											
Total conversion of carbon in feed,	* * *	0.7	10.1	18.1	18.7	16.7	7.0	2 7	Ti Special	The state of the s	
/0					-5.7	10.7	7.9	3.7	2.8	2.7	1.5
Total carbon recovery, %		0.1	0.7	2.9	5.6	8.1	14.0	24.0	0.		
							14.6	24.0	34.8	43.9	57.8
eed	5 grame	of low to-		. 1.			5, +20 U.				94.8
ime of sampling, sec.	10	20 -tell	peratur	e bitumii	nous coal	char, -16	, +20 U.	S. S sieve	e cize		
emperature ° F	1708				00	120	240	290		400	
eed hydrogen rate, SCF/hr.	98.2		1720	1720	1717	1714	1712				
XII gas rate. SCF/br	96.4	98.2	98.2	98.2	98.2	98.2	98.2	98.2		715	
xit gas composition male 07	70.4	94.9	95.3	93.9	95.7	96.0	96.7	94.9	98.2	98.2	
$N_2 + CO$	0.00							74.9	97.8	96.0	
CO_2	0.08		0.98	0.95	0.08	0.05	0.07	0.04	0.01		
H_2		0.03	0.01			0.03	0.07	0.04	0.06	0.05	
CH ₄	99.90	95.63	89.22	90.08		00 50	00.70				
C_2H_6	0.01	4.30	9.77	8.95		99.52	98.70	97.99	98.56	99.26	
n-C ₄ H ₁₀		0.01	0.01			0.43	1.22	1.96	1.38	0.68	
		0.01		* * *							
C_5H_{10}	0.01	0.01		• • •							
C_7H_{14}								V.V		0.01	
Benzene		0.01	0.01				0.01	0.01		0.01	
Total	100.00	0.01	0.01	0.02						• • •	
	100.00	100.00	100.00	100.00	100.00	100.00		100.00			
ate of formation of gaseous hydro-			11111111111111		100.00	100.00	100.00	100.00	$100.00 \ \overline{1}$	00.00	
Carbon Carbon Ib /Ib carbon										11.5	
	0.2	45 -									
ied-nr.		15.5	34.5	31.3	4.2	1 5	11	-			
ied-nr.	0.2	10.0	51.5	31.1							
otal conversion of carbon in feed	0.2			31.3	7.4	1.5	4.6	7.1	5.0	2.6	
otal conversion of carbon in feed,		1.9	5.4	10.5							
otal conversion of carbon in feed					24.8	28.5	33.8	43.2	48.2	2.6 55.5 86.4	